Pairwise Coupling Analysis of Helical Junction Hydrogen Bonding Interactions in Luteoviral RNA Pseudoknots[†]

Peter V. Cornish[‡] and David P. Giedroc*

Department of Biochemistry and Biophysics, 2128 TAMU, Texas A&M University, College Station, Texas 77843-2128

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ABSTRACT: A 28-nucleotide mRNA pseudoknot that overlaps the P1 and P2 genes of sugarcane yellow leaf virus (ScYLV) stimulates -1 ribosomal frameshifting. The in vitro frameshifting efficiency is decreased ≥8-fold upon substitution of the 3'-most loop 2 nucleotide (C27) with adenosine, which accepts a hydrogen bond from the 2'-OH group of C14 in stem S1. The solution structures of the wild-type (WT) and C27A ScYLV RNA pseudoknots show that while the RNAs adopt virtually identical overall structures, there are significant structural differences at the helical junctions of the two RNAs. Specifically, C8⁺ in loop L1 in the C8+•(G12•C28) L1-S2 major groove base triple is displaced by \approx 2.3 Å relative to the accepting stem 2 base pair (G12·C28) in the C27A RNA. Here, we use a double mutant cycle approach to analyze the pairwise coupling of the C8+•(G12•C28)···C27•(C14-G7) and ···A27•(C14-G7) hydrogen bonds in the WT and C27A ScYLV RNAs, respectively, and compare these findings with previous results from the beet western yellows virus (BWYV) RNA. We find that the pairwise coupling free energy (δ_{AB}^{i}) is favorable for the WT RNA (-0.7 ± 0.1 kcal/mol), thus revealing that formation of these two hydrogen bonds is positively cooperative. In contrast, δ_{AB}^i is 0.9 ± 0.4 kcal/mol for the poorly functional C27A ScYLV RNA, indicative of nonadditive hydrogen bond formation. These results reveal that cooperative hydrogen bond formation across the helical stem junction in H-type pseudoknots correlates with enhanced frameshift stimulation by luteoviral mRNA pseudoknots.

Intricately folded motifs in both the noncoding and coding regions of mRNAs are known to regulate all steps of the translation cycle. RNA functions in cap-independent translational initiation by providing a binding site for 40S ribosomal subunits in internal ribosomal entry sites (IRES) (1). Riboswitches are naturally occurring aptamers that bind metabolites, including adenine, guanine, S-adenosylmethionine, and thiamine pyrophosphatase, that directly regulate the expression of enzymes linked to their production or degradation (2). Folded RNA motifs found in coding regions of mRNAs regulate ribosomal recoding by stimulating stop codon suppression and stop codon redefinition or by changing the translational reading frame, an example of which is -1 programmed ribosomal frameshifting (-1 PRF) (3, 4). -1 PRF occurs during translational elongation where the mRNA slips back by one nucleotide into the "-1" reading frame relative to the reference "0" frame. In RNA viruses, -1 PRF is used to produce several proteins (P1 and P2) from a single ribosomal initiation site while regulating the relative ratios of gene products (P1 vs P1-P2) as dictated by the efficiency of the frameshift. Maintaining a proper ratio of gene products has been shown to be important for viral propagation and infectivity in retroviruses and yeast retrotransposable elements (5-9).

For efficient frameshifting to occur, several elements must be present in the mRNA (10). The first is a slip site, which in most cases conforms to the sequence X XXY YYZ, where X is any nucleotide, Y is A or U, and Z is any nucleotide except G (11). The slip site is positioned in the A and P sites of the ribosome at the time frameshifting occurs. Also required for efficient frameshifting is a RNA structural element, which in many cases is a RNA pseudoknot. This RNA structural element is separated from the 3'-edge of the slip site by a six- to eight-nucleotide linker. The linker length is likely required for optimal positioning of the pseudoknot at the mRNA entry tunnel of the ribosome where the RNA helicase activity is thought to be localized (12-14).

-1 PRF is responsible for the production of the genomically encoded P1 and P1−P2 fusion proteins in plant luteoviruses (15). The stimulatory RNA structure in luteoviral mRNAs is a hairpin (H)-type RNA pseudoknot, which contains two helical stems (S1 and S2) connected by nonequivalent loops (L1 and L2). Luteoviral RNA pseudoknots are characterized by an intricate collection of stabilizing tertiary structural loop−loop and loop−stem hydrogen bonding interactions (16−19). A C+•(G•C) L1−S2 major groove trans Watson−Crick/Hoogsteen base triple has been observed in all luteoviral pseudoknots thus far investigated, and protonation has been shown to be stabilizing by 2−3 kcal/mol at 37 °C (20). Further, this interaction is essential for

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^{*} To whom correspondence should be addressed. Phone: (979) 845-4231. Fax: (979) 845-4946. E-mail: giedroc@tamu.edu.

[‡] Current address: Department of Physics, University of Illinois, 1110 W. Green St., Urbana, IL 61801-3180.

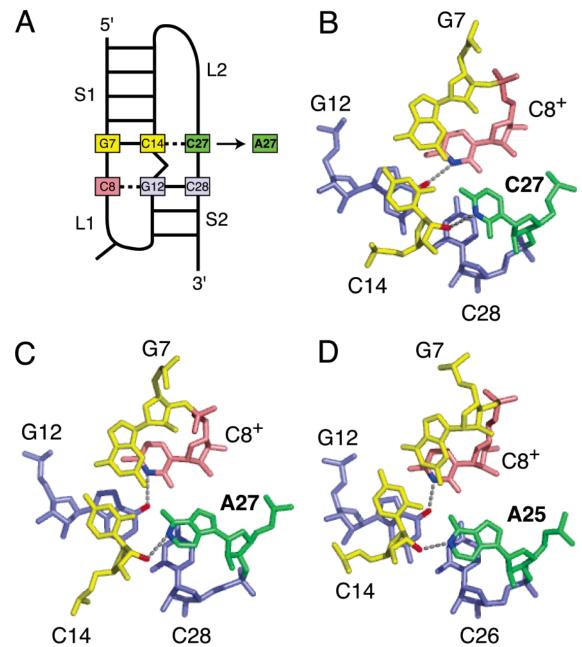


FIGURE 1: Helical junction regions of the WT ScYLV, C27A ScYLV, and BWYV frameshift-stimulating RNA pseudoknots. (A) Schematic representation of the WT ScYLV RNA pseudoknot. The noncanonical hydrogen bonding interactions investigated in this study are the trans Watson-Crick/Hoogsteen interaction between C8 (pink) and G12 (blue) and the cis Watson-Crick/sugar edge interaction between C27 (or A27) (green) and C14 (yellow). (B-D) Structures of the helical junction regions of WT ScYLV (16) (PDB entry 1YG4), C27A ScYLV (23) (PDB entry 2AP5), and WT BWYV (PDB entry 1L2X) RNA pseudoknots, respectively, with residues colored as in panel A. The dashed lines in each panel connect the donor and acceptor atoms of the two hydrogen bonds investigated in this study.

the efficient frameshift stimulation by these pseudoknots (16, 17, 21, 22). Two L2-S1 minor groove interactions at the helical junction have also been shown to be important for pseudoknot stability and maintenance of frameshifting efficiency (20).

The solution structure of the P1-P2 frameshifting mRNA pseudoknot from sugarcane yellow leaf virus (ScYLV) is unique since the L2-S1 minor groove base triple closest to the helical junction is formed by an L2 cytidine (C27) rather than an adenosine, which is found in this position in all other previously studied luteoviral RNAs (16) (see Figure 1A,B). Surprisingly, substitution of C27 with adenosine resulted in a nearly complete abrogation of ScYLV pseudoknotstimulated frameshifting in a simple in vitro assay (from 15

to 1.8%, or \geq 2-fold above slip site alone), with virtually no change in the global structure of the RNA (16, 23). The major structural difference was found at the helical junction (23). Formation of the C27·(C14·G7) minor groove triple in the WT ScYLV pseudoknot at the helical junction (Figure 1B) allows for the rearrangement of the immediately adjacent major groove C8+•(G•C) base triple on which it is stacked so that C8 moves \approx 2.3 Å relative to the accepting base pair (G12·C28) when compared to the C27A ScYLV RNA (23) (Figure 1C). This rearrangement likely contributes to a global destabilization of the C27A RNA by ≥1-1.5 kcal/mol $(\Delta \Delta G_{37})$ relative to the WT RNA (16).

The relationship between global stability and frameshift stimulation is not entirely clear. The global stability of the

downstream structural element, per se, cannot be fully responsible for frameshifting since a RNA hairpin with global stability similar to that of an RNA pseudoknot does not induce comparable levels of frameshift stimulation (24, 25). This suggests that there is some feature encoded in the pseudoknot topology itself that is capable of modulating frameshift stimulation. Since structure is unlikely to play a major role (23), one possibility is that a small localized destabilization of the helical junction itself is a major contributor to a decrease in global stability, and this, in turn, might account for functional differences between structurally similar RNA pseudoknots. We reasoned that such a destabilization of C27A ScYLV RNA relative to the WT RNA may well originate with distinct energies of pairwise coupling, δ_{AB}^{i} , of the two stacked hydrogen bonds that cross the helical junction in these two RNAs (16, 23) (see Figure 1). In this paper, we employ thermal melting experiments carried out as a function of pH with site-specific deoxycytidine (dC)-substituted RNAs to elucidate the $\delta_{AB}{}^{i}$ of the $C8^+$ H3····G12 O6 and C27 N3····C14 2'-OH (δ_{AB}^{WT}) or A27 N1···C14 2'-OH (δ_{AB}^{C27A}) hydrogen bonds at the helical junction of the WT versus C27A ScYLV RNAs and compare these findings to previous studies of the related frameshifting mRNA from beet western yellows virus (BWYV) (see Figure 1D) (20). We find that both WT RNAs are characterized by a δ_{AB}^{i} consistent with cooperative hydrogen bond formation. In contrast, the C27A ScYLV RNA is characterized by a pairwise coupling indicative of nonadditive hydrogen bond formation, despite the fact the helical junction regions of the weakly functional C27A ScYLV and wild-type BWYV RNAs are quite similar (23). The implications of these findings with respect to pseudoknot-mediated frameshift stimulation are discussed.

MATERIALS AND METHODS

RNA Synthesis and Sample Preparation. The WT and C27A ScYLV RNAs were synthesized by in vitro transcription using SP6 RNA polymerase as described previously (16). The dC14 and dC14/C27A ScYLV RNAs were purchased from Dharmacon (Boulder, CO) and deprotected according to the manufacturer's recommendations. All four RNAs were purified by denaturing PAGE, electroeluted, and desalted using C18 chromatography. The RNA samples were prepared for melting experiments by extensive dialysis (over ≥9 h) against several changes of buffer for a total of 2 L of the appropriate buffer containing 0.5 M KCl. The first buffer change contained 5 mM EDTA for the removal of adventitiously bound divalent metal ions. The following buffers were used at a final concentration of 10 mM: acetate (pH 5.0), MES (pH 5.5 and 6.0), MOPS (pH 6.5 and 7.0), EPPS (pH 7.5, 8.0, and 8.5), and CHES (pH 9.0) (26).

NMR Sample Preparation and Spectroscopy. All samples were prepared for NMR spectroscopy by multiple rounds of ethanol precipitation. The final NMR buffer consisted of 10 mM potassium phosphate, 100 mM KCl, and 5 mM MgCl₂ (pH 6.0), with a final sample concentration of \sim 2 mM for WT ScYLV, \sim 1 mM for C27A ScYLV, and \sim 0.5 mM for dC14 and dC14/C27A ScYLV RNAs. The proton resonances were referenced to an internal standard (100 μ M DSS). The jump—return echo one-dimensional spectra were recorded at a proton resonance frequency of 500 MHz. For each spectrum, 512 transients were collected with a spectral width

of 12.5 kHz and 2048 complex points. Watergate ${}^{1}H^{-1}H$ NOESY spectra ($\tau_{\rm mix} = 200$ ms) were collected with 1024 complex points in the direct dimension and 512 complex points in the indirect dimension with a spectral width of 12.5 kHz in each dimension to obtain imino proton and 2'-OH proton resonance assignments.

Optical Detection Melting Experiments and Data Analysis. Optical melting experiments were performed on a Cary 1 spectrophotometer equipped with a temperature controller. The temperature range was increased from 5 to 95 °C at a constant ramp rate of 0.3 °C/min. Absorbance data were acquired at 260 and 280 nm, and both sets of derivative data $(dA_{260}/dT \text{ and } dA_{280}/dT)$ were subjected to a simultaneous nonlinear least-squares fit to a multiple (i) sequential interacting two-state transition model that optimizes ΔH_i and $t_{\rm m,i}$ for the unfolding of each ith unfolding step as implemented by the algorithm t-melt assuming $\Delta C_p^{\circ} = 0$ (27, 28). The criteria used to determine the number of unfolding transitions are described in the text and elsewhere (16). The final RNA concentration for the UV melting experiments was 2 μ M in a 600 μ L cuvette. Three or more useable melting profiles were collected for each solution condition and analyzed individually to extract ΔH_i and $t_{m,i}$ with the resulting parameters obtained from multiple experiments averaged (see Tables S1–S4 of the Supporting Information). The unfolding entropies were obtained from the equation $\Delta S_i = \Delta H_i / t_m$, while $\Delta G_{37} = \Delta H_i - 310.15 \Delta S_i$. Folding parameters assume microscopic reversibility where ΔG_{37} (unfolding) = $-\Delta G_{37}$ (folding) and ΔH_{vH} (unfolding) = $-\Delta H_{\rm vH}$ (folding). Here, we report ΔG_{37} (folding). Melting experiments performed at higher and lower RNA concentrations were used to establish unimolecular unfolding behavior of all RNAs studied here. Additionally, extensive NMR investigation of the WT and C27A RNAs at concentrations as high as 3 mM showed no signs of aggregation (data not shown) (16).

RESULTS

Imino ¹*H NMR Spectroscopy*. It was first important to establish the degree to which the dC14 substitution influences the structure of the WT and C27A ScYLV RNAs. The downfield regions of jump-return echo one-dimensional (1D) NMR spectra acquired at 10 °C and pH 6.0 for each of the four RNAs are shown in Figure 2. The resonance assignments shown for dC14 and dC14/C27A ScYLV RNAs were confirmed by a 200 ms Watergate ¹H-¹H NOESY spectrum collected at 10 °C (data not shown). The 1D spectra of the WT and dC14 ScYLV RNAs clearly show that both RNAs are stably folded and that the C8⁺ imino and amino protons are present; both features are diagnostic for the trans Watson-Crick/Hoogsteen C+•(G•C) interaction between C8⁺ and the G12•C28 base pair (29) (see Figure 1B,C). Although the imino proton chemical shift of C8⁺ is degenerate with that of G30 in the C27A RNA and G6 in the dC14/ C27A RNA, both RNAs are stably folded since the amino protons are clearly present (Figure 1C,D). As expected, the 2'-OH group of C14 is absent in the dC14/C27A ScYLV RNA (Figure 2D). Since the WT C14 2'-OH resonance (≥ 8.5 ppm) (16) is in a crowded region of the 1D spectrum, a 200 ms NOESY spectrum was used to confirm the absence of this 2'-OH proton in the dC14 ScYLV RNA (data not shown).

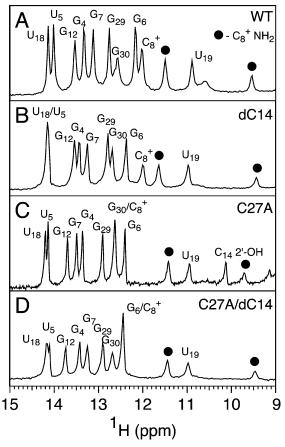


FIGURE 2: Downfield region of jump-return echo 1D NMR spectra measured at 10 °C and pH 6.0 for WT ScYLV (16) (A), dC14 ScYLV (B), C27A ScYLV (16) (C), and dC14/C27A ScYLV (D) RNAs. All four RNAs exhibit resonances for the protonated C8⁺ N3H and downfield-shifted C8⁺ NH₂ protons (●) diagnostic of the major groove trans Watson-Crick/Hoogsteen base pair (16, 17).

Scheme 1

Scheme 2

$$F \stackrel{1}{\longleftrightarrow} S1 HP \stackrel{2}{\longleftrightarrow} U$$

$$\begin{array}{c} \Delta H_1 & \Delta H_2 \\ t_{m,1} & t_{m,2} \end{array}$$

pH Dependence of the Unfolding of the WT ScYLV and Mutant RNAs. The unfolding of luteoviral RNA pseudoknots has been investigated previously by thermal unfolding as monitored by UV spectroscopy at 260 and 280 nm and differential scanning calorimetry (16, 20, 26). These RNA pseudoknots are composed of a 5 bp S1 [4 bp in the case of potato leaf roll virus (PLRV) (19)], a 3 bp S2, a twonucleotide loop L1, and a seven- to nine-nucleotide loop L2. A coupled equilibrium model comprising three sequential, two-state unfolding transitions (Scheme 1) previously established for the BWYV pseudoknot (26) was applied to other wild-type and variant pseudoknots when sufficient resolution of individual unfolding transitions was observed on the temperature coordinate (20). Each equilibrium unfolding step is characterized by a van't Hoff enthalpy of unfolding ($\Delta H_{\rm vH}$) and a melting temperature ($t_{\rm m}$) (see

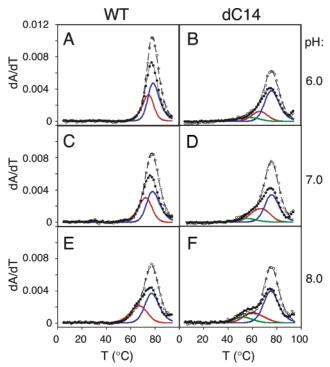


FIGURE 3: Optically monitored thermal melting profiles acquired at 260 (●) and 280 nm (○) for the WT (A, C, and E) and dC14 (B, D, and F) ScYLV RNA pseudoknots at pH 6.0 (A and B), 7.0 (C and D), and 8.0 (E and F). The smooth lines through the data represent the results of a nonlinear least-squares simultaneous fit of the A_{260} and A_{280} melting profiles to either a two-transition (A, C, and E) or three-transition (B, D, and F) unfolding model. Bestfit component unfolding transitions at A_{260} are shown as solid green, red, and blue lines as defined by the two or three sequential unfolding transitions described by Scheme 1 (red and blue lines) or 2 (green, red, and blue lines), respectively. The resulting thermodynamic parameters derived from these fits are compiled in Tables S1 and S2. Conditions: 10 mM buffer salt and 0.5 M KCl. Only every fifth data point is shown for clarity.

Schemes 1 and 2). In the three-step unfolding profile, the first unfolding step is assigned to disruption of the loopstem (L2-S1 and L1-S2) interactions in the fully folded (F) RNA pseudoknot to form PK, which is followed by the unfolding of the weaker stem S2 (S1 HP), and subsequently stem S1 (U). In cases where the first two unfolding steps were coincident on the temperature axis and sufficient unfolding enthalpy was recovered during this transition, Scheme 2 was used to analyze the melting profiles. In this scheme, the disruption of the tertiary structural interactions and S2 occurs during the first unfolding transition, giving rise to a single S1 HP intermediate (16).

Thermal melting profiles (dA/dT) monitored by UV spectroscopy were obtained for the WT, dC14, C27A, and dC14/C27A ScYLV RNAs at pH values ranging from 5.0 to 9.0 at 0.5 M KCl, the latter used to facilitate comparison with previous studies with the BWYV RNA (see Figures 3 and 4) (20). Inspection of these melting profiles reveals that the $t_{\rm m}$ values for all transitions except the last one are strongly affected by the pH; since the last unfolding transition is reporting on S1 melting, the $t_{\rm m}$ is, as expected, independent of pH (data not shown) (16). This is consistent with the unfolding models shown (Schemes 1 and 2) and with protonation of N3 of C8 only in the folded RNA (16, 20, 26). It is interesting to note that the dC14 RNAs, which destroy single C27 N3···C14 2'-OH and A27 N1···C14 2'-

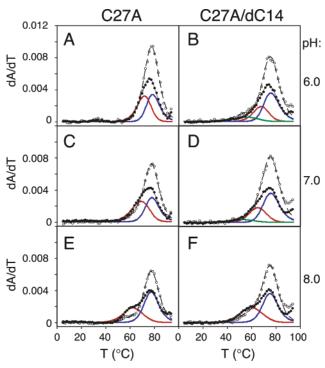


FIGURE 4: Optically monitored thermal melting profiles acquired at 260 () and 280 nm () for the C27A (A, C, and E) and dC14/C27A (B, D, and F) ScYLV RNA pseudoknots at pH 6.0 (A and B), 7.0 (C and D), and 8.0 (E and F). The smooth lines drawn through the data represent the results of a nonlinear least-squares simultaneous fit of the A_{260} and A_{280} melting profiles to either a two-transition (A, C, and E) or three-transition (B, D, and F) unfolding model. Best-fit component unfolding transitions at A_{260} are shown as the solid green, red, and blue lines as defined by the two or three sequential unfolding transitions described by Scheme 1 (red and blue) or 2 (green, red, and blue lines), respectively. The resulting thermodynamic parameters derived from these fits are compiled in Tables S3 and S4. Conditions: 10 mM buffer salt and 0.5 M KCl. Only every fifth data point is shown for clarity.

OH hydrogen bonds in WT and C27A ScYLV RNAs, respectively, break the coincident unfolding of S2 and tertiary structural interactions that characterize both WT and C27A ScYLV RNAs.

The pH dependence of folding, ΔG_{37} as calculated in Materials and Methods, is plotted in Figure 5 (\bullet) for the indicated luteoviral RNA with data analyzed by two independent pH-dependent folding models. The first model is a simple binding model (eq 1) that attributes the entire change in ΔG_{37} as a function of pH to protonation of a single group in the folded RNA according to the function

$$\begin{split} \Delta G_{37,\text{obs}} &= \Delta G_{37}^{\text{ unprotonated}} + \\ &(\Delta G_{37}^{\text{ protonated}} - \Delta G_{37}^{\text{ unprotonated}}) [10^{\text{p}K_a-\text{pH}}/(1+10^{\text{p}K_a-\text{pH}})] \end{split} \tag{1}$$

where $\Delta G_{37}^{\text{unprotonated}}$ and $\Delta G_{37}^{\text{protonated}}$ represent the ΔG_{37} values for the fully unprotonated and fully protonated state of the folded RNA molecules, respectively (20). Here, p K_a is an apparent p K_a , where K_a is equal to the acid dissociation equilibrium constant that governs C8⁺ deprotonation. This may not reflect the intrinsic p K_a , since this analysis ignores differential binding of H⁺ to the unfolded state relative to the folded state outside of this titratable group, and linkage to acid-induced or alkaline-induced unfolding (see below)

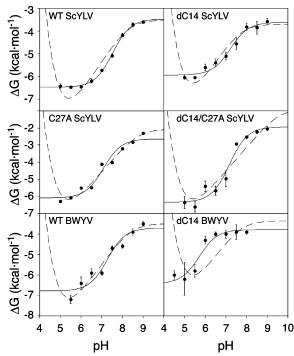


FIGURE 5: Measured ΔG_{37} values vs pH for WT, dC14, C27A, dC14/C27A, BWYV, and dC14 BWYV pseudoknots fit to two different models. $\Delta G = \Delta G_1$ for WT ScYLV RNA (see Table S1). $\Delta G = \Delta G_1 + \Delta G_2$ for dC14 RNA (see Table S2). $\Delta G = \Delta G_1$ for C27A ScYLV RNA (see Table S3). $\Delta G = \Delta G_1 + \Delta G_2$ for dC14/ C27A RNA (between pH 5 and 7). $\Delta G = \Delta G_2$ (between pH 7.5 and 9) (see Table S4). The solid line (binding equation) through the experimental data represents a fit to a binding model, which ascribes the pH dependence of the unfolding to protonation of a single titratable group on the folded RNA (see the text for details) (eq 1). The dashed line (linkage equation) corresponds to a fit to the linkage model, which ascribes the pH dependence of the unfolding to a single ionizable group in the folded state, with multiple ionizable groups in the acid-induced unfolded state (eqs 2-4). The parameter values and associated standard errors from these fits are compiled in Table 1.

(30). The fits to the data are displayed in Figure 5 (solid line) with the resolved thermodynamic parameters from the fits compiled in Table 1. As previously shown, the C27A RNA is destabilized with respect to the WT ScYLV RNA due to the rearrangement of the helical junction by \geq 1.2 kcal/mol over the entire pH range (16). $\Delta G_{37}^{\text{protonated}}$ of the dC14 RNA is 0.6 ± 0.2 kcal/mol lower than that of WT ScYLV RNA, with $\Delta G_{37}^{\text{unprotonated}}$ the same within experimental error. Interestingly, the downward shift for C27A in the apparent p K_a is only 0.4 unit, which is much smaller than the 1.6 unit shift previously observed for the dC14 BWYV RNA (20). In contrast to the dC14 ScYLV RNA, $\Delta G_{37}^{\text{protonated}}$ for dC14/C27A ScYLV RNA is approximately equal to $\Delta G_{37}^{\text{protonated}}$ for the reference C27A RNA ($\Delta\Delta G^{\text{protonated}}$ = 0.2 \pm 0.3 kcal/mol), with the $\Delta G_{37}^{\text{unprotonated}}$ more positive by 0.7 ± 0.3 kcal/mol; here, there is a negligible shift in the apparent p K_a .

The second model used to fit the ΔG_{37} versus pH data is a linkage model that accounts for linkage of a single protonation site in the folded state to acid-induced denaturation that occurs on protonation of all cytidine N3 and adenosine N1 groups in the unfolded state (30). The data were fit to the equation

$$\Delta G_{\rm f}^{\rm obs} = \Delta G_{\rm f}^{\rm ref} - RT \ln(\sum_{\rm f} / \sum_{\rm u})$$
 (2)

Table 1: Folding Free Energies (ΔG_{37}) and p K_a Values Derived from Analysis of the pH Dependence of the Unfolding of Various Luteoviral RNAsa

RNA	$\Delta G_{37}^{ ext{protonated}} ext{ (kcal/mol)}$	$\Delta G_{37}^{\text{unprotonated}}$ (kcal/mol)	$\Delta\Delta G_{37}^{\text{protonation }b}$ (kcal/mol)	pK_a	$\Delta G_{37}^{7.0}$ (kcal/mol)
WT ScYLV	-6.5 ± 0.1^{c}	-3.5 ± 0.1^{c}	-3.0 ± 0.1^{c}	7.5 ± 0.1^{c}	-5.7 ± 0.1^{c}
		-3.5 ± 0.3^d		8.3 ± 0.2^{d}	
dC14 ScYLV	-5.9 ± 0.1^{c}	-3.6 ± 0.1^{c}	-2.3 ± 0.1^{c}	7.2 ± 0.1^{c}	-5.1 ± 0.2^{c}
		-3.7 ± 0.2^{d}		7.7 ± 0.2^{d}	
C27A ScYLV	-6.1 ± 0.2^{c}	-2.6 ± 0.2^{c}	-3.5 ± 0.3^{c}	7.1 ± 0.2^{c}	-4.1 ± 0.1^{c}
		-2.1 ± 0.4^{d}		8.8 ± 0.3^{d}	
dC14/C27A ScYLV	-6.3 ± 0.2^{c}	-1.9 ± 0.3^{c}	-4.4 ± 0.4^{c}	7.2 ± 0.2^{c}	-5.0 ± 0.5^{c}
		-0.9 ± 2.9^{d}		9.6 ± 2.1^{d}	
WT $BWYV^e$	-6.8 ± 0.3^{c}	-3.7 ± 0.3^{c}	-3.1 ± 0.4^{c}	7.3 ± 0.2^{c}	-5.8 ± 0.3^{c}
		-3.5 ± 0.3^d		8.4 ± 0.3^{d}	
dC14 BWYV ^e	-6.4 ± 0.3^{c}	-3.8 ± 0.2^{c}	-2.6 ± 0.4^{c}	5.7 ± 0.2^{c}	-4.0 ± 0.2^{c}
		-3.4 ± 1.7^{d}		7.7 ± 1.4^{d}	

^a Derived from a nonlinear least-squares analysis of the melting profiles (see Figures 3 and 4) using a sequential interacting multiple-transition model (see Experimental Procedures). $^{b}\Delta\Delta G_{37}^{\text{protonation}} = \Delta G_{37}^{\text{protonated}} - \Delta G_{37}^{\text{unprotonated}}$. c Calculated from the binding model (eq 1). $^{d}\Delta G_{5}^{\text{ref}}$ (see eq 2). Calculated from the linkage model (eqs 2-4). ^e Values taken from ref 20.

where $\Delta G_{\rm f}^{\rm ref}$ is for the ligand free U to F reaction (conceptually analogous to $\Delta G_{37}^{\text{unprotonated}}$ in the binding model), and

$$\sum_{f} = 1 + 10^{pK_{a,f} - pH} \tag{3}$$

$$\sum_{\mathbf{u}} = (1 + 10^{pK_{\mathbf{a},\mathbf{u}}^A - pH})^{n^A} (1 + 10^{pK_{\mathbf{a},\mathbf{u}}^C - pH})^{n^C}$$
 (4)

where n^A and n^C equal the number of adenosines and cytidines in the molecule, respectively. The pK_a values for $pK_{a,u}^{A}$ and $pK_{a,u}^{C}$ were fixed to 3.7 and 4.4, respectively, and represent the intrinsic pKa of N1 of adenosine and N3 of cytosine, respectively (30). This leaves just two fitted parameters, $pK_{a,f}$ and ΔG_f^{ref} . The two terms Σ_f and Σ_u reflect the sum contributions of protonation in the folded and unfolded states of the RNA, respectively. Best fit curves to the linkage equation are superimposed (dashed lines) on the experimental data (Figure 5) with the p $K_{\rm a,f}$, and $\Delta G_{\rm f}^{\rm ref}$ values given in Table 1. As can be seen, $\Delta G_{37}^{\rm unprotonated}$ and $\Delta G_{\rm f}^{\rm ref}$ are in reasonably good agreement for all of the RNA molecules. Note that the calculated pK_a values predicted from the linkage model are significantly higher than those from the binding equation, as expected (30). The linkage equation should better resolve the microscopic pK_a for protonation of C8⁺ in the folded state, provided data are obtained at pH < 5.0, since this is when the two functions strongly diverge from one another; however, this point was not investigated here (30).

Effect of pH on the Unfolding of dC14 RNAs. To gain insight into the effect of pH on the unfolding of the dC14 RNAs relative to the WT and C27A RNA counterparts, $\Delta\Delta G_{37}$ was calculated for the binding model (eq 5) and linkage model (eq 6) and the simulated curves, derived from the parameter values compiled in Table 1, superimposed on the experimental data, $\Delta\Delta G_{37}^{\rm exp}$ (Figure 6).

$$\Delta \Delta G_{37,\text{obs}}(\text{pH}) = \Delta G_{37,\text{obs}}^{\text{deoxyC14}}(\text{pH}) - \Delta G_{37,\text{obs}}^{\text{WT}}(\text{pH})Z$$
 (5)

$$\Delta\Delta G_{\rm f}^{\rm obs}({\rm pH}) = \Delta G_{\rm f}^{\rm obs, deoxyC14}({\rm pH}) - \Delta G_{\rm f}^{\rm obs,WT}({\rm pH}) \quad (6)$$

Although there is significant scatter given the small $\Delta\Delta G_{37}$ values, the simulated curves generated from the binding model capture the general trends in the experimental data reasonably well (Figure 6A). In contrast, the simulated curve generated from the linkage model approximates the data less

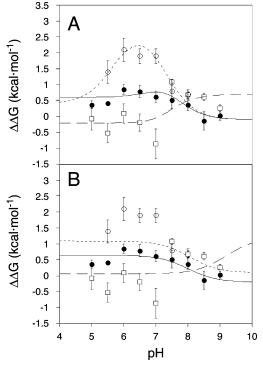


FIGURE 6: Influence of pH on dC14 substitution mutants vs the corresponding WT counterpart. (A and B) Experimentally determined values: (**(•)** $\Delta\Delta G_{37} = \Delta G_{37}^{\text{dC14}} - \Delta G_{37}^{\text{WT}}$, (**(□)** $\Delta\Delta G_{37} = \Delta G_{37}^{\text{C27A/dC14}} - \Delta G_{37}^{\text{C27A/dC14}}$, and (**(○)** $\Delta\Delta G_{37} = \Delta G_{37}^{\text{BWYV/dC14}} - \Delta G_{37}^{\text{BWYV/dC14}}$ ΔG_{37}^{BWYV} (20) from the data shown in Figure 5. (A) Simulated curves derived from the parameters from the binding equation fits (eq 1 and Table 1) to the pH dependence: (—) $\Delta\Delta G_{37}(\text{pH}) = \Delta G_{37}^{\text{dC14}}(\text{pH}) - \Delta G_{37}^{\text{WT}}(\text{pH}), (---) \Delta\Delta G_{37}(\text{pH}) = \Delta G_{37}^{\text{C27A/dC14}}(\text{pH}) - \Delta G_{37}^{\text{C27A/dC14}}(\text{pH}), \text{ and } (---) \Delta\Delta G_{37}(\text{pH}) = \Delta G_{37}^{\text{BWYV/dC14}}(\text{pH}) - \Delta G_{37}^{\text{BWYV}}(\text{pH}).$ (B) Simulated curves derived from the parameters obtained from the linkage equation fits to the pH dependence (eqs 2-4): (-) $\Delta\Delta G_f(pH) = \Delta G_f^{dC14}(pH)$ - $\Delta G_{\rm f}^{\rm WT}(\rm pH), (---) \Delta \Delta G_{\rm f}(\rm pH) = \Delta G_{\rm f}^{\rm C27A/dC14}(\rm pH) - \Delta G_{\rm f}^{\rm C27A}(\rm pH),$ and (---) $\Delta \Delta G_f(pH) = \Delta G_f^{BWYV/dC14}(pH) - \Delta G_f^{BWYV}(pH)$.

well (Figure 6B). The reason for this is embodied in the function (eq 2) itself which reduces to

$$\Delta\Delta G_{\rm f}^{\rm obs}({\rm pH}) = \Delta\Delta G_{\rm f}^{\rm ref} - RT \ln(\sum_{\rm f}^{\rm mut}/\sum_{\rm f}^{\rm WT}) ~~(7)$$

where $\Delta\Delta G_{\rm f}^{\rm ref} = \Delta G_{\rm f}^{\rm ref,mut} - \Delta G_{\rm f}^{\rm ref,WT}$ provided $\Sigma_{\rm u}^{\rm WT} =$ $\Sigma_{\rm u}^{\rm mut}$. This is a reasonable assumption since the two RNAs are the same length and have identical base

Table 2: Pairwise Coupling Free Energies (37 °C) Derived from the Binding Equation for the Six-Component Thermodynamic Cycles from the Thermodynamic Cube Shown in Figure 7^a

reference state	perturbation A/B	$\Delta G_{ m A}{}^{ m i}$ (kcal/mol)	$\Delta G_{\rm B}{}^{\rm i}$ (kcal/mol)	$\Delta G_{\mathrm{AB}^{\mathrm{i}}} (\mathrm{kcal/mol})$	$\delta_{{\sf AB}^{{\sf i}b}}$ (kcal/mol)
WT•H ⁺	$dC14/-H^+$ (δ_{AB}^{WT} face)	0.6 ± 0.1	3.0 ± 0.1	2.9 ± 0.1	$-0.7 \pm 0.1 (\delta_{AB}^{WT})$
C27A•H ⁺	$dC14/-H^+$ (δ_{AB}^{C27A} face)	-0.2 ± 0.3	3.5 ± 0.3	4.2 ± 0.2	$0.9 \pm 0.4 (\delta_{AB}^{C27A})$
$WT \cdot H^+$	C27A/dC14 (δ_{AB}^{C8+} face)	0.4 ± 0.2	0.6 ± 0.1	$0.2 \pm 0.v$	$-0.8 \pm 0.3 (\delta_{AB}^{C8^+})$
WT	C27A/dC14 (δ_{AB}^{C8} face)	0.9 ± 0.2	-0.1 ± 0.4	1.6 ± 0.3	$0.8 \pm 0.4 (\delta_{AB}^{C8})$
$WT \cdot H^+$	C27A/-H ⁺ (front face)	0.4 ± 0.2	3.0 ± 0.1	3.9 ± 0.3	$0.5 \pm 0.4 (\delta_{AB}{}^{Y})$
dC14•H ⁺	C27A/-H+ (back face)	-0.4 ± 0.2	2.3 ± 0.1	4.0 ± 0.3	$2.1 \pm 0.4 (\delta_{AB}^{Z})$
$BWYV \cdot H^{+c}$	dC14/-H+	0.4 ± 0.4	3.1 ± 0.4	3.0 ± 0.4	$-0.5 \pm 0.7 (\delta_{AB}^{BWYV})$

^a Derived from $\Delta G^{\text{protonated}}$ and $\Delta G^{\text{deprotonated}}$ for the WT, dC14, C27A, and dC14/C27A ScYLV RNA pseudoknots as described in the text. ^b $\delta_{AB}{}^{i}$ = $\Delta G_{AB}{}^{i}$ – ($\Delta G_{A}{}^{i}$ + $\Delta G_{B}{}^{i}$). Note that conservation of free energy requires that $\delta_{AB}{}^{WT}$ – $\delta_{AB}{}^{C27A}$ = $\delta_{AB}{}^{C8}{}^{+}$ – $\delta_{AB}{}^{C8}$ = $\delta_{AB}{}^{Y}$ – $\delta_{AB}{}^{Z}$ = -1.6 kcal/mol. ^c $\Delta G_{J}{}^{i}$ values taken from ref 20.

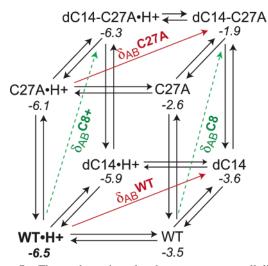


FIGURE 7: Thermodynamic cube that encompasses all linkage relationships between the various ScYLV RNAs investigated in this work. The global reference state is the wild-type ScYLV RNA at low pH (WT·H⁺). The italic numbers below each RNA represent the folding free energies (ΔG_{37}) derived from the binding equation (eq 1) fits (given in Table 1). Each δ_{AB}^i is described in the text, with all six pairwise δ_{AB}^i values compiled in Table 2.

compositions (the same number of A and C residues) and differ only by a single oxygen atom. When mutant and WT RNAs have identical pK_a values, the second term in eq 7 vanishes, reducing to $\Delta\Delta G_f^{\text{obs}}(\text{pH}) = \Delta\Delta G_f^{\text{ref}}$, i.e., a straight line. In the case where $pK_a^{\text{mut}} \neq pK_a^{\text{WT}}$, the function $\Delta\Delta G_f^{\text{obs}}(\text{pH})$ is strictly monotonic, i.e., always increasing or decreasing.

Pairwise Coupling, δ_{AB}^i , across the Helical Junction in WT versus C27A ScYLV Pseudoknots. A standard double mutant cycle strategy was next employed to obtain δ_{AB}^{WT} and δ_{AB}^{C27A} using the WT·H⁺ as the overall system reference state (see Figure 7). δ_{AB}^{WT} is the pairwise coupling free energy that defines the extent to which elimination of both C27 N3···C14 2′-OH (via the deoxy-C14 substitution) and C8⁺ H3···G12 O6 (via an increase in pH) hydrogen bonds in the WT ScYLV RNA (ΔG_{AB}^{WT}) differs from the sum of the loss of each hydrogen bond individually (ΔG_A^{WT} and ΔG_B^{WT} , respectively) (31). Likewise, δ_{AB}^{C27A} defines the analogous free energy difference for the C27A ScYLV pseudoknot, i.e., loss of the A27 N1···C14 2′-OH (ΔG_A^{C27A}) and C8⁺ H3···G12 O6 (ΔG_B^{C27A}) hydrogen bonds individually or in combination (ΔG_{AB}^{C27A}). δ_{AB}^i is therefore defined as

$$\delta_{AB}^{i} = \Delta G_{AB}^{i} - (\Delta G_{A}^{i} + \Delta G_{B}^{i}) \tag{8}$$

 $\Delta G_{\rm A}{}^{\rm i},\,\Delta G_{\rm B}{}^{\rm i},$ and $\Delta G_{\rm AB}{}^{\rm i}$ are obtained by manipulation of $\Delta G_{37}^{\text{protonated}}$ for WT·H⁺, dC14·H⁺, C27A·H⁺, and dC14/ C27A·H⁺ ScYLV RNAs, i.e., the RNA species on the left side of the thermodynamic cube (Figure 7), and $\Delta G_{37}^{\text{unprotonated}}$ for the same four deprotonated RNAs, i.e., those on the right side of the thermodynamic cube (Figure 7). These $\Delta G_{37}^{\text{protonated}}$ and $\Delta G_{37}^{\text{unprotonated}}$ values are compiled in Table 1 and reproduced on the thermodynamic cube in Figure 7 (in italics). For example, $\Delta G_{\rm A}^{\rm WT} = -5.9 - -6.5 = 0.6$ kcal/ mol, $\Delta G_{\rm B}^{\rm WT} = -3.5 - -6.5 = 3.0$ kcal/mol, and $\Delta G_{\rm AB}^{\rm WT}$ = -3.6 - -6.5 = 2.9 kcal/mol. Thus, $\delta_{AB}^{WT} = 2.9 - (3.0)$ + 0.6) = -0.7 kcal/mol, reflecting a favorable pairwise coupling free energy and indicative of positive cooperativity of hydrogen bond formation ($\delta_{AB}^{i} \le 0$) (see Table 2). Note that if $\delta_{AB}^{i} = 0$, then $\Delta G_{AB}^{i} = \Delta G_{A}^{i} + \Delta G_{B}^{i}$. In this case, formation of each hydrogen bond is additive or noncooperative, and the loss of the A or B interaction is not influenced by the loss of the remaining interaction at site B or A, respectively. If $\delta_{AB}^{i} > 0$, the pairwise coupling is unfavorable and formation of both hydrogen bonds is nonadditive. $\Delta G_{\rm A}^{\rm i}$, $\Delta G_{\rm B}{}^{\rm i}, \ \Delta G_{\rm AB}{}^{\rm i}, \ \delta_{\rm AB}{}^{\rm WT}, \ {\rm and} \ \delta_{\rm AB}{}^{\rm C27A}$ values calculated in this way are compiled in Table 2, as are the corresponding values obtained from a similar analysis of the closely related BWYV and dC14 BWYV pseudoknots published previously (20). Interestingly, both δ_{AB}^{WT} and δ_{AB}^{BWYV} are significantly negative, indicative of positively cooperative formation of these two hydrogen bonds that straddle the helical junction. In contrast, δ_{AB}^{C27A} has a positive sign, indicative of nonadditivity of hydrogen bond formation across the helical junction. This finding is consistent with the free energy cycles that yield $\delta_{AB}^{C8^+}$ and δ_{AB}^{C8} (Figure 7, left and right). $\delta_{AB}^{C8^+}$ and δ_{AB}^{C8} are equal to -0.8 and 0.8 kcal/mol, respectively (see Table 2), and reflect the extent to which perturbation of the H-bond donor (dC14) and acceptor (C27A) sides of the same C27 N3···C14 2'-OH hydrogen bond, individually and in combination, is influenced by protonation of immediately adjacent C8. At low pH, this coupling is favorable $(\delta_{AB}^{C8^+} < 0)$; i.e., the folding free energy of the double mutant is less negative than the sum of each of the underlying single mutants, with the opposite true at high pH in the deprotonated forms of the RNA ($\delta_{AB}^{C8} > 0$). Finally, using the WT·H+ conformation as the reference state for the thermodynamic cube, and denoting $\Delta G_{\rm C}^{\rm WT}$ as $\Delta \Delta G^{\rm protonated}$ for the C27A versus WT ScYLV RNAs and $\Delta G_{
m ABC}^{
m WT}$ as $\Delta\Delta G^{\text{protonated}}$ for the dC14/C27A versus WT RNAs (and $\Delta G_{\rm A}^{
m WT}$ and $\Delta G_{\rm B}^{
m WT}$ defined as above), one can also calculate the three-way coupling in this system, ΔG^3 , according to

$$\Delta G^{3} = \Delta G_{\text{ABC}}^{\text{WT}} - \left[(\Delta G_{\text{A}}^{\text{WT}} + \Delta G_{\text{B}}^{\text{WT}} + \Delta G_{\text{C}}^{\text{WT}}) + \sum \delta_{\text{AB}}^{\text{i}} \right] (9)$$

where ΔG^3 is equal to the folding free energy of the triple substitution mutant dC14/C27A RNA at high pH, $\Delta G_{\rm ABC}^{\rm WT}$, minus the sum of the folding free energies of the single mutants and each of the six pairwise couplings, $\delta_{\rm AB}{}^{\rm i}$ (Table 2). ΔG^3 is -1.8 kcal/mol, which is indicative of a strong coupling network in the WT ScYLV RNA.

Effect of pH on $\delta_{AB}{}^{WT}$, $\delta_{AB}{}^{C27A}$, and $\delta_{AB}{}^{BWYV}$ Values. This coupling analysis is completely independent of p K_a since the states selected were derived from $\Delta G^{\text{protonated}}$ and $\Delta G^{\text{deprotonated}}$ measured at low and high pH, respectively. To understand the effect of pH on the three pairwise couplings, $\delta_{AB}{}^{\text{i}}$ ($\delta_{AB}{}^{\text{WT}}$, $\delta_{AB}{}^{\text{C27A}}$, and $\delta_{AB}{}^{\text{BWYV}}$), the coupling scheme was modified to correct for the concentration of reference state molecules, i.e., WT•H⁺, C27A•H⁺, or BWYV•H⁺, at each pH. Here, the equations change to

$$\Delta G_{\rm AB}^{\rm eff}(\rm pH) = \Delta G_{37}^{\rm unprotonated, deoxyC14} [\rm HA]^{\rm deoxyC14} - \\ [\Delta G_{37,\rm obs}^{\rm WT}(\rm pH)] (\rm [HA]^{\rm WT}) \ \, (10)$$

$$\Delta G_{\rm A}^{\rm eff}({\rm pH}) = [\Delta G_{\rm 37,obs}^{\rm deoxyC14}({\rm pH})][{\rm HA}]^{\rm deoxyC14} - \\ \Delta G_{\rm 37,obs}^{\rm WT}({\rm pH})([{\rm HA}]^{\rm WT}) \ (11)$$

$$\Delta G_{\rm B}^{\rm eff}({\rm pH}) = [\Delta G_{37}^{\rm unprotonated,WT} - \Delta G_{37,{\rm obs}}^{\rm WT}({\rm pH})]({\rm [HA]}^{\rm WT})$$
(12)

$$[HA^{X}] = 1/(1 + 10^{pH - pK_{a}^{X}})$$
 (13)

for the binding model, allowing us to define a new term, $\delta_{AB}^{i,eff}(pH)$, the effective pairwise coupling as a function of pH, according to

$$\begin{split} \delta_{\rm AB}^{\quad i,\rm eff}(\rm pH) &= \Delta G_{\rm AB}^{\quad i,\rm eff}(\rm pH) - \\ & [\Delta G_{\rm A}^{\quad i,\rm eff}(\rm pH) + \Delta G_{\rm B}^{\quad i,\rm eff}(\rm pH)] \end{split} \tag{14}$$

Exactly analogous expressions were derived for the linkage model (see eqs 2-4), using p K_a and ΔG_f^{ref} parameters resolved from the linkage model fits (Table 1, italicized parameters). Plots of $\delta_{AB}{}^{i,eff}$ as a function of pH are presented as simulated curves using parameters derived from the binding model (Figure 8A) and linkage model (Figure 8B). As expected, at low pH, $\delta_{AB}^{i,eff}(pH) = \delta_{AB}^{i}$ (from Table 2) for each RNA; this occurs as a result of taking the WT·H⁺, C27A·H+, or BWYV·H+ state as the reference state in each case. In Figure 8A, $\delta_{AB}^{i,eff}$ first becomes more negative or favorable for the WT ScYLV (solid line) and BWYV (dotted line) RNAs; the depth of this trough is essentially determined by the difference in binding pK_a values between the WT and dC14-containing RNAs, which is greater for the BWYV RNA (see Table 1) (20). As the pH is further increased, $\delta_{AB}^{i,eff}$ becomes less negative and vanishes altogether at pH \geq 8.0, fully expected since the concentration of protonated RNA molecules becomes very small at high pH. The situation is dramatically different for the C27A ScYLV RNA, where $\delta_{AB}^{i,eff}$ is positive or unfavorable at all pH values, until vanishing as expected at high pH (Figure 8A, dashed line). Roughly similar trends in the simulated curves are obtained with the linkage model-derived parameters (Figure 8B),

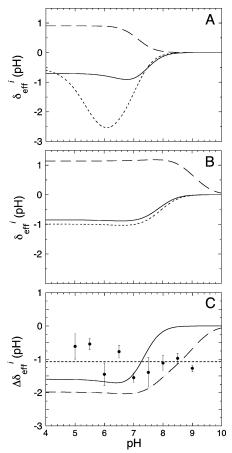


FIGURE 8: Effective coupling free energies, $\delta_{AB}^{i,eff}$, as a function of pH. (A) Simulated curves derived from eq 14 are shown for the WT ScYLV (—), C27A ScYLV (——), and BWYV (---) RNAs derived from parameters obtained from the binding equation fits (Table 1) to eqs 10–13. (B) Simulated curves derived from eq 14 for parameters derived from the linkage equation fits (Table 1). (C) Simulated curves drawn for $\Delta\delta_{AB}^{eff} = \delta_{AB}^{WT,eff} - \delta_{AB}^{C27A,eff}$ derived for the binding equation fits (solid line, which equals the solid line minus the dashed line from panel A) and linkage equation fits (dashed line, which equals the solid line minus the dashed line in panel B), compared to global $\Delta\Delta G_{37}$ values measured previously (16). Dotted line, average global $\Delta\Delta G_{37}$.

except that there are no troughs since eq 7 is monotonic, and $\delta_{AB}^{i,eff}$ vanishes at significantly higher pH values due to the higher p K_a values derived from this analysis (Table 1).

Figure 8C shows the results of plotting $\Delta \delta_{AB}^{i,eff}$, defined as $\delta_{AB}^{WT} - \delta_{AB}^{C27A}$, derived from the binding (solid line) and linkage (dashed line) model parameters compared to the experimental global $\Delta\Delta G_{37}$ measured as a function of pH (16). This is a measure of the degree to which the relative differences in pairwise coupling free energies across the helical junction measured for the wild-type and C27A RNAs track with global differences in stability between the two RNAs. Although the trends in $\Delta\Delta G_{37}$ do not show a simple pH dependence, the average value of $\Delta\Delta G_{37}$ (-1.1 \pm 0.3 kcal/mol, dotted line) has the same sign and magnitude and is well within the simulated values of $\Delta \delta_{AB}{}^{i,eff}$ over the entire pH range, with $\Delta \delta_{AB}^{i,eff}$ being more negative than $\Delta \Delta G_{37}$ at pH <7 and less negative at higher pH values. This is particularly so considering a conservative estimate of ± 0.5 kcal/mol uncertainty in the experimental values of $\Delta \delta_{\rm eff}$ particularly at low pH (see Table 2). These data are therefore generally consistent with the idea that much of the global $\Delta\Delta G_{37}$ can be traced to an altered pairwise coupling of these two cross junction hydrogen bonds between functional (WT) and poorly functional (C27A) ScYLV RNA pseudoknots. This is fully consistent with the structures of these two RNAs (16, 23).

DISCUSSION

In this paper, we have determined the extent to which two key adjacent tertiary structural hydrogen bonds that cross the helical junction form cooperatively in three H-type RNA pseudoknots for which structures and frameshift stimulation activities (at pH 7.0) are known. The most striking finding from this study is that in the context of the two most conservative substitutions that can be introduced, i.e., the loss of the H-bond donor in each of the two hydrogen bonds (C14 2'-OH···2'-H and C8 N3-H+···N3) individually or in combination, the WT ScYLV (δ_{AB}^{WT}) and WT BWYV (δ_{AB}^{BWYV}) RNAs exhibit favorable pairwise coupling free energies (Table 2) indicative of cooperative hydrogen bond formation, while the poorly functional C27A ScYLV RNA (δ_{AB}^{C27A}) is characterized by an unfavorable coupling, or nonadditive hydrogen bond formation. This is the case despite the fact that the helical junction region of the C27A ScYLV RNA is more like that of the BWYV RNA relative to the wild-type ScYLV RNA (see Figure 1) (23).

This result reveals that the energetics of pairwise hydrogen bond coupling is not predicted well by the global structure alone, but it is instead strongly influenced by the context in which these interactions are made. An implicit assumption in any pairwise coupling experiment is that the structures of the molecule lacking one or both hydrogen bonds are essentially identical and all that is missing is the hydrogen bond itself and the energetic consequences that derive from deleting it. Since hydrogen bond strength is influenced by hydrogen bond geometry (length and angle), the extent to which formation of two hydrogen bonds is pairwise coupled could also be influenced by small structural differences among all four RNAs at low and high pH. Very high resolution structures of these RNAs under a variety of solution conditions would be required to address this. In fact, introduction of a single deoxyribose substitution in an otherwise all-RNA context may also perturb the sugar pucker from C3'-endo (N-type) to C2'-endo (S-type) (32), albeit to varying degrees, which might indirectly influence the magnitude of hydrogen bond coupling by perturbing the local structure. However, this is clearly not the case here since NMR data reveal that the dC14 sugar pucker adopts a C3'endo conformation, like C14 in the WT RNA (16). In general, the measured pairwise coupling free energy may not be direct, and the presence or absence of specific interactions in the BWYV versus C27A ScYLV RNAs may play a role in determining the magnitude and sign of the pairwise coupling free energy. One way to establish whether a measured pairwise coupling free energy is direct or indirect is to introduce a mutation or substitution into a third site and to re-evaluate δ_{AB}^{i} (31). If the two values of δ_{AB}^{i} are not equal, then the AB pairwise coupling is indirect, thus providing evidence of a more complex network of thermodynamically coupled or cooperative interactions. In this regard, it is important to further consider the structural differences between the C27A ScYLV and BWYV RNAs. Comparison of panels C and D of Figure 1 reveals that the helical junction of the C27A ScYLV RNA is more strongly over-rotated than the BWYV RNA, which results in poorer stacking of the two junction triple base pairs; this, in turn, might influence the degree of pairwise coupling of junction hydrogen bonds in these two RNAs. However, this cannot be the sole reason for the different δ_{AB}^{i} values since the WT ScYLV RNA is even more poorly stacked than in the C27A RNA (see Figure 1B,C).

The other structural difference between the BWYV and C27A ScYLV RNAs is that the ScYLV pseudoknot contains an extrahelical nucleotide, C25, which is not present in the BWYV RNA. The presence or absence of this nucleotide might influence the magnitude of δ_{AB}^{i} via a next nearest neighbor influence on the packing of A26 into the minor groove triplex, as well as the adjacent A24 (G6 C15) base triple; clearly, the integrity of the nucleotide analogous to A26 in the ScYLV pseudoknot is important for luteoviral pseudoknot stability (19, 20, 33). Interestingly, deletion of C25 in ScYLV RNAs consistently enhances frameshift stimulation by ≥ 2 -fold in a variety of structural contexts; simultaneous deletion of C25 with one other L2 nucleotide enhances frameshift stimulation by the C27A RNA 3-fold, resulting in a functional activity indistinguishable from that of the wild-type BWYV RNA (16). If the magnitude and sign of δ_{AB}^{i} track with frameshifting activity, this leads to the prediction that the Δ C25 and this BWYV-like ScYLV mutant would yield very different δ_{AB}^{i} values relative to those of the C27A and WT ScYLV pseudoknots.

Upon encountering a programmed mRNA frameshift signal, the decoding ribosome is thought to pause and shift reading frames without dissociation and continue decoding in the new reading frame. The pseudoknot has to be sufficiently stable to resist mechanical unwinding (34) by the elongating ribosome in a kinetically relevant rate, but not too stable to induce dissociation of the translation elongation complex. In this way of thinking, the pseudoknot provides a kinetic barrier to ribosome-mediated unfolding (and/or refolding). The studies reported here extend our previous structural work on this system and support the hypothesis that those helical junctions with favorable δ_{AB}^{i} values may well resist ribosome-mediated resolution of the pseudoknot more effectively than those characterized by less favorable or unfavorable δ_{AB}^{i} values since these would likely lower the kinetic barrier to ribosome-mediated unwinding. In this regard, it will be interesting to investigate hydrogen bond coupling throughout the molecule. For example, although deletion of the 2'-OH protons of C15 or A16 riboses might be predicted to have a significant impact on global stability (20), it would be interesting to determine how the magnitude of δ_{AB}^{i} is affected by deoxynucleotide substitutions farther from the C8+•(G12•C28) major groove base triple. This would provide insight into the cooperativity of the loop-stem interactions in frameshifting RNAs. Finally, we note that the magnitude of the pairwise coupling as

 $^{^1}$ A NOESY spectrum ($\tau_{\rm m}=120$ ms) acquired in D_2O was used to assign the H1' (5.67 ppm), H2' (2.70 ppm), and H2" (3.36 ppm) protons of dC14. Only the sequential dC14 H2'–C15 H6 NOE is clearly observed in this spectrum with the dC14 H2"–C15 H6 NOE not readily detected. In addition, there are no detectable dC14 H1'–H2' or H1'–H2" correlations in the TOCSY spectrum under conditions where sugars with known C2'-endo puckers give intense correlations, e.g., U19 (spectra not shown). All of these data are consistent with the conclusion that the dC14 sugar adopts a C3'-endo conformation in the dC14 RNA.

operationally defined here is dependent on pH, leading to the prediction that the efficiency of frameshifting induced by luteoviral pseudoknots might also be influenced by the pH. Experiments along these lines are currently in progress.

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SUPPORTING INFORMATION AVAILABLE

WT ScYLV thermodynamic parameters (Table S1), dC14 ScYLV thermodynamic parameters (Table S2), C27A ScYLV thermodynamic parameters (Table S3), and dC14/C27A ScYLV thermodynamic parameters (Table S4). This material is available free of charge via the Internet at http://pubs.acs.org.

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